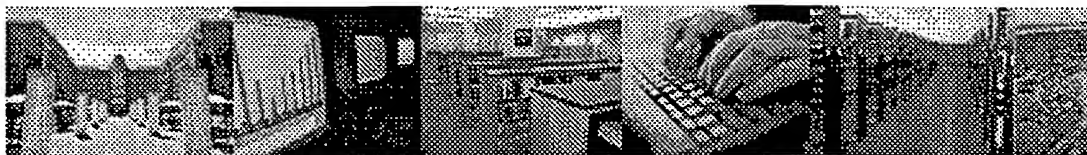




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* Country Code:

* Publication Date:

* Language:

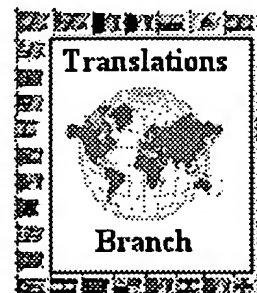
First Inventor Name:

2. ☐ Article

* Author:

* Language:

* Country:



3. ☐ Other

* Type of Document:

* Language:

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Last modified 09/27/2006 11:48:25

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FILE 'REGISTRY' ENTERED AT 09:54:32 ON 27 SEP 2006

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L3 12 S L1 FUL CSS

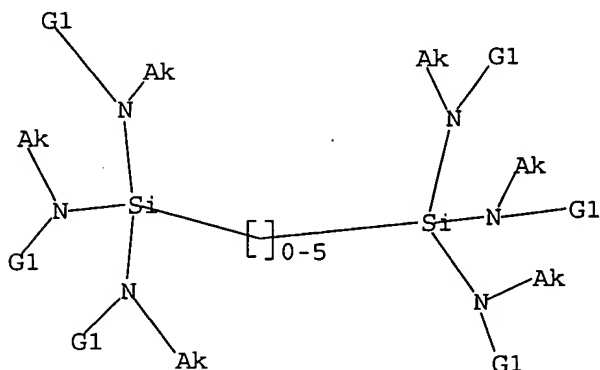
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L4 19 S L3

=> d l1

L1 HAS NO ANSWERS

L1 STR



G1 H,Ak

Structure attributes must be viewed using STN Express query preparation.

=> d bib abs hitstr 1-19

L4 ANSWER 1 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2006:317164 CAPLUS

DN 144:362341

TI pH stable chromatographic media using templated multilayer
organic/inorganic grafting

IN Rustamov, Ismail M.; Chitty, Michael C.; Farkas, Tivadar; Loo, Lawrence;
Welch, Emmet

PA USA

SO U.S. Pat. Appl. Publ., 13 pp.

CODEN: USXXCO

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2006070937	A1	20060406	US 2005-240695	20050930
	WO 2006039507	A2	20060413	WO 2005-US35217	20050930
	WO 2006039507	A3	20060908		
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG,				

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YU, ZA, ZM, ZW
RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
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CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,
GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
KG, KZ, MD, RU, TJ, TM

PRAI US 2004-615093P P 20041001

US 2004-615812P P 20041004

AB An advanced silica gel sorbent for use in chromatog. sepns. that was chemical modified by surface polycondensation of a trifunctional and/or difunctional organosilane. The chromatog. media exhibits a wider pH range and improved pH stability as compared to other silica gel based sorbents, while retaining all other pos. aspects attributed to silica gel based sorbents. A method of forming the advanced silica gel sorbent by Templated Multilayer Inorg./Organic Grafting.

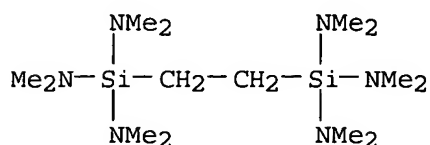
IT 20248-45-7

RL: RCT (Reactant); RACT (Reactant or reagent)

(pH stable chromatog. stationary phase using templated multilayer organic/inorg. grafting)

RN 20248-45-7 CAPLUS

CN 2,7-Diaza-3,6-disilaooctane-3,3,6,6-tetramine,
N,N,N',N',N'',N'',N''',N''',2,7-decamethyl- (9CI) (CA INDEX NAME)



L4 ANSWER 2 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2005:429719 CAPLUS

DN 142:472926

TI Low temperature deposition of silicon nitride

IN Senzaki, Yoshihide; Helms, Aubrey L.

PA Aviza Technology, Inc., USA

SO PCT Int. Appl., 14 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005045899	A2	20050519	WO 2004-US36018	20041029
	WO 2005045899	A3	20060302		
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RW:	BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	US 2005227017	A1	20051013	US 2004-976697	20041028
	EP 1682692	A2	20060726	EP 2004-796762	20041029

CAS ONLINE PRINTOUT

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, HR

PRAI US 2003-518608P P 20031031
US 2004-976697 A 20041028
WO 2004-US36018 W 20041029

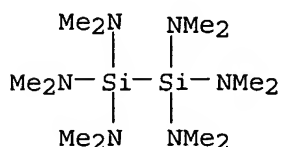
OS MARPAT 142:472926

AB A novel class of volatile liquid precursors based on amino substituted
disilane compds. was used to form Si nitride dielec. materials on the
surface of substrates. This class of precursors overcomes the issues of
high deposition temps. and the formation of undesirable byproducts that
are inherent in the present art. In another aspect, methods of depositing
Si nitride films on substrates are provided.

IT 6415-17-4P
RL: CPS (Chemical process); PEP (Physical, engineering or chemical
process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)
(preparation and low temperature deposition of silicon nitride using
volatile liquid
precursors based on amino substituted disilane compds.)

RN 6415-17-4 CAPLUS

CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)



L4 ANSWER 3 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:1037401 CAPLUS

DN 142:14033

TI CVD method for forming silicon nitride film

IN Kato, Hitoshi; Fukushima, Kohei; Yonezawa, Masato; Hiraga, Junya

PA Tokyo Electron Limited, Japan

SO PCT Int. Appl., 47 pp.
CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2004105115	A1	20041202	WO 2004-JP7311	20040521
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	JP 2005012168	A2	20050113	JP 2004-45508	20040220
PRAI	JP 2003-148332	A	20030526		
	JP 2004-45508	A	20040220		
AB	A CVD method for forming a Si nitride film comprises a step where while exhausting air from a process chamber in which a substrate to be processed is placed, a silane gas and NH3 gas are supplied into the chamber and a Si				

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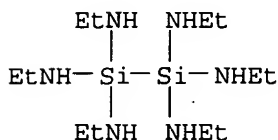
nitride film is formed on the substrate by CVD. This Si nitride film-forming step comprises a 1st period during when the silane gas is supplied into the process chamber and a 2nd period during when the supply of the silane gas is suspended, and the 1st period alternates with the 2nd period.

IT 532980-53-3

RL: NUU (Other use, unclassified); USES (Uses)
(CVD of silicon nitride film)

RN 532980-53-3 CAPLUS

CN Disilanehexamine, N,N',N'',N''',N''',N''''-hexaethyl- (9CI) (CA INDEX NAME)



RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 4 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:569892 CAPLUS

DN 141:106612

TI Preparation of amino substituted disilane derivatives for composition and method for low temperature deposition of silicon-containing films

IN Wang, Ziyun; Xu, Chongying; Baum, Thomas H.; Hendrix, Bryan; Roeder, Jeffrey F.

PA USA

SO U.S. Pat. Appl. Publ., 9 pp., Cont.-in-part of U.S. Ser. No. 294,431.
CODEN: USXXCO

DT Patent

LA English

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2004138489	A1	20040715	US 2003-699079	20031031
	US 2004096582	A1	20040520	US 2002-294431	20021114
	WO 2004044958	A2	20040527	WO 2003-US36097	20031112
	WO 2004044958	A3	20040826		
	W:				
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	AU 2003287710	A1	20040603	AU 2003-287710	20031112
	EP 1567531	A2	20050831	EP 2003-781915	20031112
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	JP 2006517517	T2	20060727	JP 2004-552143	20031112
PRAI	US 2002-294431	A2	20021114		
	US 2003-699079	A	20031031		
	WO 2003-US36097	W	20031112		
OS	CASREACT 141:106612; MARPAT 141:106612				

CAS ONLINE PRINTOUT

AB This invention relates to silicon precursor compns. for forming silicon-containing films by low temperature (e.g., <300°) chemical vapor deposition processes for fabrication of ULSI devices and device structures. Such silicon precursor compns. comprise at least one disilane derivative compound that is fully substituted with alkylamino and/or dialkylamino functional groups. Thus, amination of (Et₂N)₂(Cl)SiSi(Cl)(NEt₂) with Me₂NH in Et₂O gave 90% (Et₂N)(NHMe)SiSi(NHMe)(NEt₂)₂ which was used as silicon precursor for silicon-containing films.

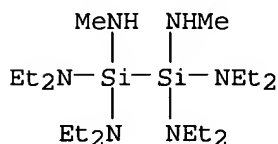
IT 693827-57-5P 693827-58-6P

RL: NUU (Other use, unclassified); RCT (Reactant); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(preparation of amino substituted disilane derivs. for composition and method for low temperature deposition of silicon-containing films)

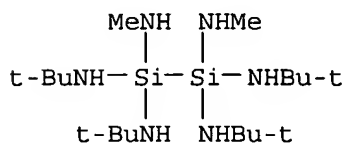
RN 693827-57-5 CAPLUS

CN Disilanehexamine, N₁,N₁',N₁',N₁',N₂,N₂,N₂'N₂'-octaethyl-N₁'',N₂''-dimethyl- (9CI) (CA INDEX NAME)



RN 693827-58-6 CAPLUS

CN Disilanehexamine, N₁,N₁',N₂,N₂'-tetrakis(1,1-dimethylethyl)-N₁'',N₂''-dimethyl- (9CI) (CA INDEX NAME)



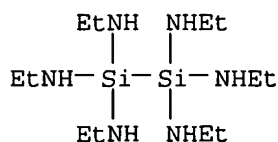
IT 532980-53-3

RL: NUU (Other use, unclassified); RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)

(preparation of amino substituted disilane derivs. for composition and method for low temperature deposition of silicon-containing films)

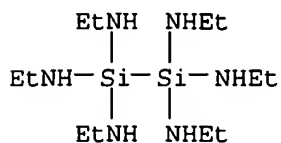
RN 532980-53-3 CAPLUS

CN Disilanehexamine, N,N',N'',N''',N''',N''''-hexaethyl- (9CI) (CA INDEX NAME)

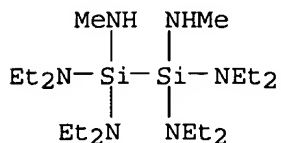


DN	141:15676					
TI	Composition and method for low temperature deposition of silicon-containing films such as films including silicon, silicon nitride, silicon dioxide and/or silicon oxynitride					
IN	Wang, Ziyun; Xu, Chongying; Laxman, Ravi K.; Baum, Thomas H.; Hendrix, Bryan; Roeder, Jeffrey					
PA	Advanced Technology Materials, Inc., USA					
SO	PCT Int. Appl., 69 pp. CODEN: PIXXD2					
DT	Patent					
LA	English					
FAN.CNT	3					
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
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PI	WO 2004044958	A2	20040527	WO 2003-US36097	20031112	
	WO 2004044958	A3	20040826			
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	US 2004138489	A1	20040715	US 2003-699079	20031031	
	AU 2003287710	A1	20040603	AU 2003-287710	20031112	
	EP 1567531	A2	20050831	EP 2003-781915	20031112	
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK					
	JP 2006517517	T2	20060727	JP 2004-552143	20031112	
PRAI	US 2002-294431	A	20021114			
	US 2003-699079	A	20031031			
	WO 2003-US36097	W	20031112			
OS	MARPAT 141:15676					
AB	Si precursors for forming Si-containing films in the manufacture of semiconductor devices, such as low dielec. constant (k) thin films, high k gate silicates, low temperature Si epitaxial films, and films containing Si nitride (Si3N4), silicon oxynitride (SiOxNy) and/or SiO2. The precursors of the invention are amenable to use in low temperature (e.g., < 500° or <300°) CVD processes, for fabrication of ULSI devices and device structures.					
IT	532980-53-3P, Disilanehexamine, N,N',N'',N''',N''''',N''''''-hexaethyl- 693827-57-5P 693827-58-6P					
	RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)					
	(vapor deposition precursor; composition and method for low temperature deposition of silicon-containing films such as films including silicon, silicon nitride, silicon dioxide and/or silicon oxynitride)					
RN	532980-53-3 CAPLUS					
CN	Disilanehexamine, N,N',N'',N''',N''''',N''''''-hexaethyl- (9CI) (CA INDEX NAME)					

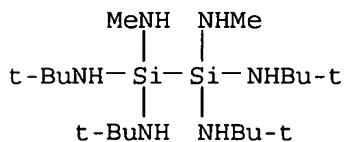
CAS ONLINE PRINTOUT



RN 693827-57-5 CAPLUS

CN Disilanehexamine, N1,N1,N1',N1',N2,N2,N2'N2''-octaethyl-N1'',N2''-dimethyl-
(9CI) (CA INDEX NAME)

RN 693827-58-6 CAPLUS

CN Disilanehexamine, N1,N1',N2,N2''-tetrakis(1,1-dimethylethyl)-N1'',N2''-
dimethyl- (9CI) (CA INDEX NAME)

L4 ANSWER 6 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:412601 CAPLUS

DN 140:432729

TI Method and precursor compounds for the low temperature deposition of
silicon-containing filmsIN Wang, Ziyun; Xu, Chongying; Laxman, Ravi K.; Baum, Thomas H.; Hendrix,
Bryan; Roeder, Jeffrey

PA USA

SO U.S. Pat. Appl. Publ., 20 pp.

CODEN: USXXCO

DT Patent

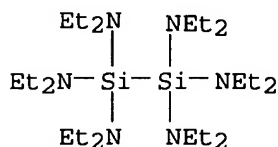
LA English

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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	US 2004138489	A1	20040715	US 2003-699079	20031031
	WO 2004044958	A2	20040527	WO 2003-US36097	20031112
	WO 2004044958	A3	20040826		
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CAS ONLINE PRINTOUT

AU 2003287710 A1 20040603 AU 2003-287710 20031112
 EP 1567531 A2 20050831 EP 2003-781915 20031112
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
 IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
 JP 2006517517 T2 20060727 JP 2004-552143 20031112
 PRAI US 2002-294431 A2 20021114
 US 2003-699079 A 20031031
 WO 2003-US36097 W 20031112
 OS MARPAT 140:432729
 AB The invention relates to a method and precursor compds. for the low temperature
 deposition of silicon-containing films, such that the films are more easily
 deposited with tight geometric features and reduced feature size. The
 silicon-containing films include low dielec. constant thin films, high-k gate
 silicates, low temperature silicon epitaxial films, and films containing
 silicon
 nitride (Si₃N₄), siliconoxynitride (SiO_xN_y) and/or silicon dioxide (SiO₂).
 The precursors of the invention are amenable to use in low temperature
 (<500°) chemical vapor deposition processes, for fabrication of ULSI
 devices and device structures.
 IT 145700-17-0
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (vapor deposition precursor; method and precursor compds. for low
 temperature
 deposition of silicon-containing films)
 RN 145700-17-0 CAPLUS
 CN Disilanehexamine, dodecaethyl- (9CI) (CA INDEX NAME)



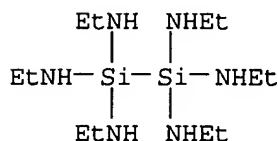
L4 ANSWER 7 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
 AN 2003:434792 CAPLUS
 DN 139:15272
 TI Method for depositing silicon nitride films and silicon oxynitride films
 by chemical vapor deposition
 IN Dussarrat, Christian; Girard, Jean-Marc
 PA Air Liquide SA pour l'Etude et l'Exploitation des Procedes Georges Claude,
 Fr.
 SO PCT Int. Appl., 23 pp.
 CODEN: PIXXD2
 DT Patent
 LA English
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2003046253	A1	20030605	WO 2002-EP13869	20021127
	W:				
	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,				
	CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,				
	GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,				
	LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,				
	PL, PT, RO, RU, SC, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT,				
	TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW:				
	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,				
	KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES,				
	FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF,				

CAS ONLINE PRINTOUT

CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG

JP 2003168683 A2 20030613 JP 2001-367126 20011130
 AU 2002356634 A1 20030610 AU 2002-356634 20021127
 EP 1458903 A1 20040922 EP 2002-803815 20021127
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
 IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK
 US 2005037627 A1 20050217 US 2004-497455 20041012
 US 6936548 B2 20050830
 PRAI JP 2001-367126 A 20011130
 WO 2002-EP13869 W 20021127
 OS MARPAT 139:15272
 AB This invention describes a method for the production of Si nitride and Si oxynitride films by CVD technol., wherein said method provides acceptable film deposition rates even at lower temps. and is not accompanied by the production of large amts. of NH₄Cl. Use of a hydrocarbylaminodisilane compound (R₀)₃-Si-Si-(R₀)₃ {each R₀ is independently selected from the hydrogen atom, chlorine atom, and -NR₁(R₂) groups (wherein R₁ and R₂ are each independently selected from the hydrogen atom and C₁ to C₄ hydrocarbyl with the proviso that R₁ and R₂ may not both be the hydrogen atom) and at least one R₀ is the -NR₁(R₂) group} as a precursor for Si nitride and Si oxynitride.
 IT 532980-53-3P
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (precursor; synthesis and use of hexakis(ethylamino)disilane in depositing silicon nitride films and silicon oxynitride films by CVD)
 RN 532980-53-3 CAPLUS
 CN Disilanehexamine, N,N',N'',N''',N''',N''''-hexaethyl- (9CI) (CA INDEX NAME)



RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 8 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
 AN 2003:434568 CAPLUS
 DN 139:28880
 TI Hexakis(monohydrocarbylamino) disilanes and method for the preparation thereof
 IN Dussarrat, Christian; Girard, Jean-Marc
 PA Air Liquide SA pour l'Etude et l'Exploitation des Procèdes Georges Claude, Fr.
 SO PCT Int. Appl., 13 pp.
 CODEN: PIXXD2
 DT Patent
 LA English
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2003045959	A1	20030605	WO 2002-EP13790	20021127
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,				

CAS ONLINE PRINTOUT

PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ,
 UA, UG, US, UZ, VN, YU, ZA, ZM, ZW
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
 KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES,
 FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF,
 CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG

JP 2003171383 A2 20030620 JP 2001-367123 20011130
 AU 2002361979 A1 20030610 AU 2002-361979 20021127
 EP 1458730 A1 20040922 EP 2002-796575 20021127
 EP 1458730 B1 20060503

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
 IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK

CN 1592750 A 20050309 CN 2002-823445 20021127
 AT 325127 E 20060615 AT 2002-796575 20021127
 US 2005107627 A1 20050519 US 2004-497399 20041227
 US 7019159 B2 20060328
 US 2006030724 A1 20060209 US 2005-222361 20050908
 US 7064083 B2 20060620

PRAI JP 2001-367123 A 20011130
 WO 2002-EP13790 W 20021127
 US 2004-497399 A1 20041227

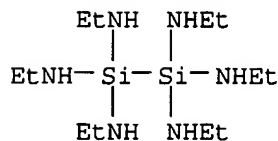
OS MARPAT 139:28880

AB This invention describes silane compds. that are free of chlorine, provide excellent film-forming characteristics at low temps. in the case of Si nitride films and Si oxynitride films, and also have excellent handling characteristics. This invention also provides a method for preparing these silane compds. which are hexakis (monohydrocarbylamino) dislanes ((R)HN)3-Si-Si-(NH(R))3 wherein each R independently represents C1 to C4 hydrocarbyl. These disilanes can be synthesized by reacting hexachlorodisilane in organic solvent with at least 6-fold moles of the monohydrocarbylamine RNH2 (wherein R is C1 to C4 hydrocarbyl).

IT 532980-53-3P, Hexakis(hydroethylamino) disilane
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (preparation of hexakis(monohydrocarbylamino) disilanes for use in CVD of Si nitride films and Si oxynitride films)

RN 532980-53-3 CAPLUS

CN Disilanehexamine, N,N',N'',N''',N''',N''''-hexaethyl- (9CI) (CA INDEX NAME)



RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 9 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
 AN 2003:398432 CAPLUS
 DN 138:394066
 TI Formation of oxide films for semiconductor devices
 IN Machida, Hideaki; Shimoyama, Norio
 PA Tri Chemical Laboratory Inc., Japan
 SO Jpn. Kokai Tokkyo Koho, 7 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese

CAS ONLINE PRINTOUT

FAN.CNT 1

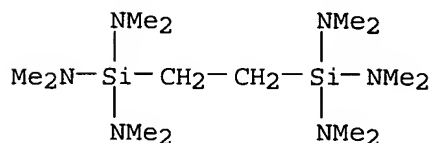
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2003151972	A2	20030523	JP 2001-350486	20011115
PRAI	JP 2001-350486		20011115		

AB Si oxide type films containing at least Si, O, C and H are formed through the dissoln. of ≥ 1 Si type compds. $R_nSi(OR)_{4-n}$ ($R = H$, alkyl, alkoxide or amino group; $n = 0, 1, 2, 3$ or 4) and $R_3Si(CH_2)_mSiR_3$ ($R = H$, alkyl, alkoxide or amino group; R_s may be different; and $m = \text{integer} \geq 1$), and polymerization of monomers. The oxide films thus formed are suited for insulation of Cu interconnections of semiconductor devices.

IT 20248-45-7 527707-21-7
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (dissoln. of silicon compds. and polymerization of monomers in formation of oxide films for semiconductor devices)

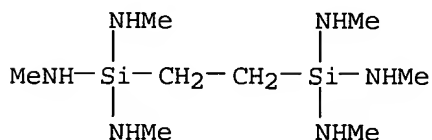
RN 20248-45-7 CAPLUS

CN 2,7-Diaza-3,6-disilaoctane-3,3,6,6-tetramine,
 N,N,N',N',N'',N'',N''',N''',2,7-decamethyl- (9CI) (CA INDEX NAME)



RN 527707-21-7 CAPLUS

CN Silanetriamine, 1,1'-(1,2-ethanediyl)bis[N,N',N''-trimethyl- (9CI) (CA INDEX NAME)



L4 ANSWER 10 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2002:407300 CAPLUS

DN 136:410026

TI Materials and method for forming Si-type insulator films for semiconductor devices

IN Machida, Hideaki; Noda, Naoto

PA Tri Chemical Laboratory Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.
 CODEN: JKXXAF

DT Patent

LA Japanese

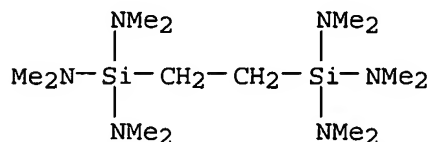
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2002158223	A2	20020531	JP 2000-350528	20001117
PRAI	JP 2000-350528		20001117		

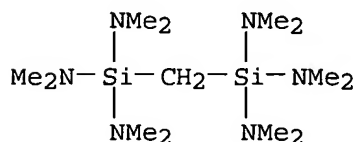
AB The insulator film are formed using Si-type materials with the formula: $\{R_3(R_4)N\}_3Si-\{C(R_1)R_2\}_n-Si\{N(R_5)R_6\}_3$, where $R_1, R_2 = H$, hydrocarbon groups, or X(halogen atom)-substituted hydrocarbon groups (R_1 and R_2 can be same), $n = 1-5$ integer, R_3, R_4, R_5 and $R_6 = H$, hydrocarbon groups or X(halogen atom)-substituted hydrocarbon groups (R_3, R_4, R_5 and R_6 can be

CAS ONLINE PRINTOUT

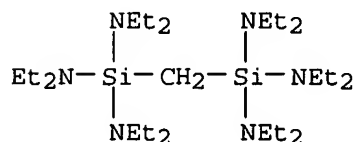
same). The insulator films may be formed on substrates by CVD.
 IT 20248-45-7 75738-28-2 431949-49-4
 431949-50-7
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (materials and method for forming Si-type insulator films for
 semiconductor devices)
 RN 20248-45-7 CAPLUS
 CN 2,7-Diaza-3,6-disilaooctane-3,3,6,6-tetramine,
 N,N,N',N',N'',N'',N''',N''',2,7-decamethyl- (9CI) (CA INDEX NAME)



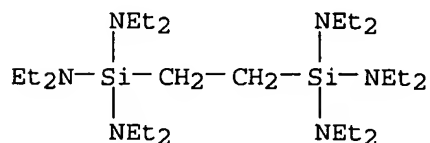
RN 75738-28-2 CAPLUS
 CN 2,6-Diaza-3,5-disilaheptane-3,3,5,5-tetramine,
 N,N,N',N',N'',N'',N''',N''',2,6-decamethyl- (9CI) (CA INDEX NAME)



RN 431949-49-4 CAPLUS
 CN Silanetriamine, 1,1'-methylenebis[N,N,N',N',N'',N''-hexaethyl- (9CI) (CA INDEX NAME)



RN 431949-50-7 CAPLUS
 CN Silanetriamine, 1,1'-(1,2-ethanediyl)bis[N,N,N',N',N'',N''-hexaethyl- (9CI) (CA INDEX NAME)



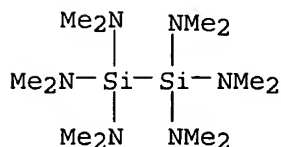
L4 ANSWER 11 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
 AN 2000:94721 CAPLUS
 DN 132:237123
 TI Disilane-Catalyzed and Thermally Induced Oligomerizations of Alkynes: A
 Comparison
 AU Yang, Jinchao; Verkade, John G.

CAS ONLINE PRINTOUT

CS Department of Chemistry, Iowa State University, Ames, IA, 50011, USA
 SO Organometallics (2000), 19(5), 893-900
 CODEN: ORGND7; ISSN: 0276-7333
 PB American Chemical Society
 DT Journal
 LA English
 AB The alkynes RC.tplbond.CR (R = H, Et, Ph), RC.tplbond.CH (R = Me(CH₂)₅, Me(CH₂)₇, Ph, Me₃Si, EtO₂C), and RC.tplbond.CR' (R = Ph, R' = C₆F₅; R = Me, R' = Ph) trimerize to corresponding benzene derivs. in 30-100% yields in the presence of Si₂Cl₆ as a procatalyst at 170-200° over 20-48 h. These reactions represent only the 2nd example of nonmetal-catalyzed alkyne trimerizations. The unsym. alkynes Me₃SiC.tplbond.CH, EtO₂CC.tplbond.CH, and PhC.tplbond.CC₆F₅ gave sym. 1,3,5-substituted benzenes, while the others led to isomeric mixts. A 1:2 M mixture of PhC.tplbond.CH and PhC.tplbond.CPh provided an isomeric mixture (45% yield) consisting mainly of 1,2,4,5-tetraphenylbenzene. While Si₂(OMe)₆ also catalyzed alkyne trimerizations (though not as efficiently as Si₂Cl₆), Si₂Me₆ did not, suggesting an electronegativity influence in the formation of the Cl₃Si• radicals shown to be involved in these reactions. Somewhat unexpectedly, however, neither Si₂F₆ nor sym-Si₂Me₂Cl₄ catalyzed alkyne trimerizations. Exptl. support for the radical pathway proposed for the alkyne trimerization observed herein is presented. In the absence of disilane procatalyst, PhC.tplbond.CH gave an isomeric mixture of dimers, p-MeC₆H₄C.tplbond.CH afforded predominantly a single dimer, and 1-ethynyl-1-cyclohexene provided exclusively a single dimer, whereas RC.tplbond.CH (R = alkyl) and PhC.tplbond.CMe did not react upon heating under the same conditions.

IT 6415-17-4, Hexakis(dimethylamino)disilane
 RL: CAT (Catalyst use); USES (Uses)
 (thermally induced trimerization of alkynes to give benzene derivs. catalyzed by)

RN 6415-17-4 CAPLUS
 CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)



RE.CNT 43 THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 12 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
 AN 1997:442988 CAPLUS
 DN 127:161886
 TI Preparation and characterization of the carbosilazanes
 bis[tris(methylamino)silyl]methane and bis[tris(phenylamino)silyl]methane
 AU Jansen, M.; Bzik, S.
 CS Institut Anorganische Chemie, Universitat Bonn, Bonn, D-53121, Germany
 SO Zeitschrift fuer Naturforschung, B: Chemical Sciences (1997), 52(6),
 707-710
 CODEN: ZNBSEN; ISSN: 0932-0776
 PB Verlag der Zeitschrift fuer Naturforschung
 DT Journal
 LA German
 AB [(RNH)₃Si]₂CH₂ (R = Me, Ph) were synthesized as potential precursors of porous O-free solids by the reaction of (Cl₃Si)₂CH₂ with MeNH₂ and with lithiated aniline, resp. [(PhNH)₃Si]₂CH₂ was characterized by crystal

CAS ONLINE PRINTOUT

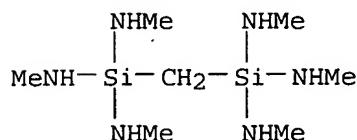
structure anal. It crystallizes in the monoclinic space group P21/c with a 10.963(2), b 17.801(2), c 17.557(2) Å, β 97.96(2)°, and Z = 4 (R1 = 4.4%, wR2 = 9.8%, 5950 independent reflections).

IT 193748-19-5P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of bis[tris(organoamino)silyl]methanes)

RN 193748-19-5 CAPLUS

CN 2,6-Diaza-3,5-disilaheptane-3,3,5,5-tetramine, N,N',N'',N'''-tetramethyl-
(9CI) (CA INDEX NAME)



L4 ANSWER 13 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1996:212092 CAPLUS

DN 124:276075

TI Manufacture of silicon nitride-based electrically insulating film by
plasma CVD

IN Kito, Hideyoshi

PA Sony Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 08022986	A2	19960123	JP 1994-153855	19940705
PRAI	JP 1994-153855		19940705		

AB The title method involves successive formation of (1) a SiN-based or SiON-based underlayer elec. insulating thin film with relatively high amount of hydrocarbon groups from a reactant gas containing an organic Si compound with

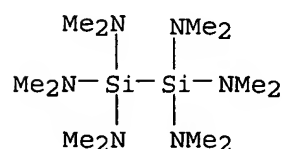
Si-N linkage and (2) a SiN-based overlayer elec. insulating film with relatively low amount of hydrocarbon groups on a substrate by CVD. The film is useful as a passivation film or an interlayer insulating film in semiconductor devices. The film was formed with improved step coverage and showed good water resistance.

IT 6415-17-4, Hexakis(dimethylamino)disilane

RL: RCT (Reactant); RACT (Reactant or reagent)
(reactant gas; manufacture of silicon nitride-based elec. insulating film by plasma CVD)

RN 6415-17-4 CAPLUS

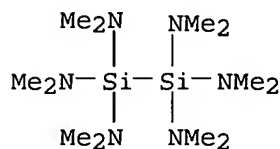
CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)



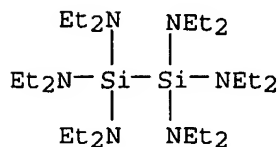
L4 ANSWER 14 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

CAS ONLINE PRINTOUT

AN 1993:72469 CAPLUS
DN 118:72469
TI Synthesis of (dialkylamino)disilanes
AU Wan, Yanjian; Verkade, John G.
CS Dep. Chem., Iowa State Univ., Ames, IA, 50011, USA
SO Inorganic Chemistry (1993), 32(3), 341-4
CODEN: INOCAJ; ISSN: 0020-1669
DT Journal
LA English
AB The previous preparation of (Me₂N)₃SiSi(NMe₂)₃ (1) (E. Wiberg, et al., 1965) was stated to proceed quant. The present preparation repeatedly gave a mixture of only .apprx.40% 1 and 60% of (Me₂N)₃SiSi(NMe₂)₂Cl (2). 1 Was made in 84% yield by treating the aforementioned mixture with LiNMe₂ in THF, and 2 can be prepared in 91% yield from Si₂Cl₆ and excess HNMe₂ using Et₂O as the solvent. Preps. are reported for (Me₂N)₃SiSi(NMe)₂OMe, (Et₂N)₃SiSi(NEt₂)₃, and (Me₂N)₃SiOSi(NMe₂)₃. The possible role of steric hindrance in the complete substitution of Cl groups in Si₂Cl₆ by NR₂ moieties is discussed. Crystal data: 1; monoclinic, space group P2₁/c, a 9.563(1), b 13.765(1), c 8.515(9) Å, α 90.0, β 115.313(8), γ 90.0°, Z = 2, R = 0.038, R_w = 0.053. The structural metrics give some indication of steric compression of the substituents around the waist of the mol.
IT 6415-17-4P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(preparation and crystal structure and reactions of, with triethanolamine or tris(aminoethyl)amine)
RN 6415-17-4 CAPLUS
CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)



IT 145700-17-0P
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)
RN 145700-17-0 CAPLUS
CN Disilanehexamine, dodecaethyl- (9CI) (CA INDEX NAME)



L4 ANSWER 15 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN
AN 1987:496884 CAPLUS
DN 107:96884
TI Process for the preparation of olefinic silanes and siloxanes
IN Quirk, Jennifer M.; Kanner, Bernard
PA Union Carbide Corp., USA
SO U.S., 8 pp.
CODEN: USXXAM

CAS ONLINE PRINTOUT

DT Patent
LA English
FAN.CNT 1

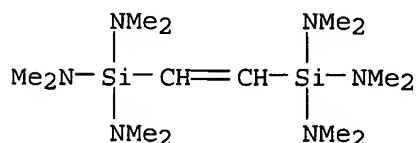
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 4668812	A	19870526	US 1985-815003	19851231
	CA 1290762	A1	19911015	CA 1986-525896	19861219
	BR 8606482	A	19871020	BR 1986-6482	19861229
	AU 8667047	A1	19870702	AU 1986-67047	19861230
	AU 598780	B2	19900705		
	EP 228095	A2	19870708	EP 1986-118123	19861230
	EP 228095	A3	19880803		
	EP 228095	B1	19920122		
	R: CH, DE, FR, GB, IT, LI, NL, SE				
	JP 62164688	A2	19870721	JP 1986-315976	19861230
	JP 03053317	B4	19910814		
PRAI	US 1985-815003	A	19851231		

AB The title compds. are prepared by dehydrogenative silylation of olefins in the presence of Rh or Ru catalysts. A mixture of 25 g (Me₂N)₃SiH and 0.95 mg RhCl₂(CO)₄ in xylene was autoclaved with ethylene at 50° and 1200 psi. Heating was continued to 148° and 1450 psi where an exotherm occurred to 225° and 1900 psi. At this point the reaction was cooled giving 87.4% CH₂:CHSi(NMe₂)₃ and 10.6% EtSi(NMe₂)₃. A variety of olefins and silanes and siloxanes were tried.

IT 109706-02-7P
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of, by dehydrogenative silylation of ethylene)

RN 109706-02-7 CAPLUS

CN 2,7-Diaza-3,6-disila-oct-4-ene-3,3,6,6-tetramine,
N,N,N',N',N'',N'',N''',N''',2,7-decamethyl- (9CI) (CA INDEX NAME)



L4 ANSWER 16 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1981:15823 CAPLUS

DN 94:15823

TI Germatranes. II. Synthesis of (triorganylsilylmethyl)germatranes

AU Gar, T. K.; Khromova, N. Yu.; Nosova, V. M.; Mironov, V. F.

CS USSR

SO Zhurnal Obshchei Khimii (1980), 50(8), 1764-7

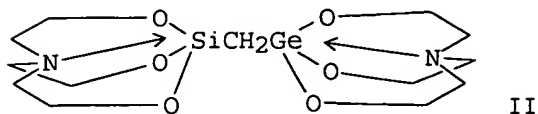
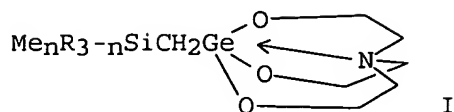
CODEN: ZOKHA4; ISSN: 0044-460X

DT Journal

LA Russian

OS CASREACT 94:15823

GI



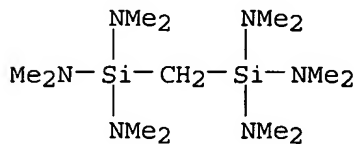
AB Cyclization of $\text{MenR}_3\text{-nSiCH}_2\text{Ge(OR}_1\text{)}_3$ ($\text{R} = \text{EtO, Me}_2\text{CHO, ClCH}_2$; $\text{R}_1 = \text{Et, Me}_2\text{CH}$; $n = 0\text{-}3$) with $\text{N(CH}_2\text{CH}_2\text{OH)}_3$ in absence of base gave 43-91% I. Similar cyclization of $(\text{Me}_2\text{N})_3\text{SiCH}_2\text{Ge(NMe}_2\text{)}_3$ with $\text{N(CH}_2\text{CH}_2\text{OH)}_3$ gave 31% II.

IT 75738-28-2P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 75738-28-2 CAPLUS

CN 2,6-Diaza-3,5-disilaheptane-3,3,5,5-tetramine,
 $\text{N,N,N',N',N'',N'',N''',N''',2,6-decamethyl- (9CI) (CA INDEX NAME)}$



L4 ANSWER 17 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1969:439145 CAPLUS

DN 71:39145

TI Organic silicon-nitrogen compounds

IN Creamer, Charles E.

PA Union Carbide Corp.

SO Ger. Offen., 45 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 1800968		19690430	DE 1968-1800968	19681003
	FR 1582475			FR	
	GB 1195159			GB	
	US 3467686		19690916	US	19671003
PRAI	US		19671003		

AB The title compds. are prepared by treating at temps. $>50^\circ$ an organosilicon compound containing at least one Si-Cl bond with an equimolar amount of an organic base containing at least one N-H bond in the presence of approx. a stoichiometric amount Mg, Ca, or Zn, and a contact time not greater than the reaction rate of the metal with the HCl or the HCl salt (I) of the base. This process avoids the formation of a troublesome and voluminous precipitate of I. Thus, to a stirred mixture of 5021 g. $\text{Cl(SiMe}_2\text{O)}_4\text{01SiMe}_2\text{Cl}$ and 300 g. Mg turnings at 100° is introduced 100 g. anhydrous Me_2NH in such a manner

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that the formation of the $\text{Me}_2\text{NH}\cdot\text{HCl}$ is observed as a slightly turbidity and a slight temperature rise is maintained. Under these conditions the temperature

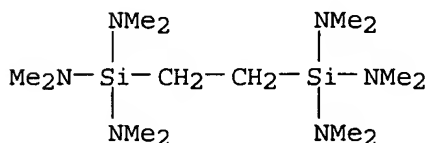
risers to 113° within 1.5 hrs. and decreases to 71° after an addnl. 3 hrs. The mixture is heated 2 hrs. at 100° to remove the turbidity caused by traces of $\text{Me}_2\text{NH}\cdot\text{HCl}$, cooled, and filtered or decanted from precipitated MgCl_2 (0.95 l.) to give 84% $\text{Me}_2\text{N}(\text{SiMe}_2\text{O})_4\cdot 0.1\text{SiMe}_2\text{NMe}_2$. Similarly are prepared 92% $\text{Me}_2\text{PhSiNMe}_2$, b2-3 53° ; 81.5% $[(\text{Me}_2\text{N})_3\text{SiCH}_2]_2$; 67% $\text{CH}_2:\text{CHSi}[\text{N}(\text{Pr-iso})_2]_3$; 90.5% $\text{Me}_2\text{Si}(\text{NMe}_2)_2$, b. 128° ; and 80.5% $\text{Me}_3\text{SiNMe}_2$, b. 82° . The compds. are useful as hydrophobic agents, intermediates for the preparation of resins, polysiloxane elastomers, and additives for lubricants and glues.

IT 20248-45-7P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 20248-45-7 CAPLUS

CN 2,7-Diaza-3,6-disilaooctane-3,3,6,6-tetramine,
 $\text{N,N,N',N',N'',N'',N''',N''',2,7-decamethyl- (9CI) (CA INDEX NAME)}$



L4 ANSWER 18 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1966:67233 CAPLUS

DN 64:67233

OREF 64:12533b-c

TI Hexadimethylaminodisilane $\text{Si}_2(\text{NMe}_2)_6$

AU Wiberg, Egon; Stecher, Oskar; Neumaier, Alfons

CS Univ. Munich, Germany

SO Inorg. Nucl. Chem. Letters (1965), 1(2), 33-4

DT Journal

LA German

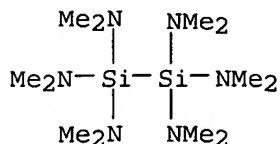
AB The title compound (I) is prepared by decomposition of Si_2Cl_6 with excess dimethylamine at room temperature, by extracting with ether and by sublimation of the

extract at 10^{-4} mm. and 70 to 80° . In damp air, I hydrolyzes slowly and is soluble in acids with decomposition to form Si_2Cl_6 , SiCl_4 , HSiCl_3 , and $\text{HNMe}_2\cdot\text{HCl}$. With alkalis, I is neither soluble nor decomposed. Its disproportionation into $\text{Si}(\text{NMe})_4$ and $\text{Si}(\text{NMe}_2)_2$ are discussed.

IT 6415-17-4, Disilanehexamine, dodecamethyl-
(preparation of)

RN 6415-17-4 CAPLUS

CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)

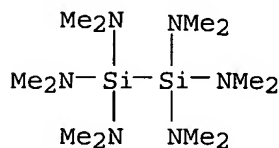


L4 ANSWER 19 OF 19 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1966:67232 CAPLUS

CAS ONLINE PRINTOUT

DN 64:67232
 OREF 64:12533a-b
 TI The autoxidation of tetrakis(dimethylamino)ethylene
 AU Urry, W. H.; Sheeto, J.
 CS Univ. of Chicago
 SO Photochemistry and Photobiology (1965), 4(6), 1067-83
 CODEN: PHCBAP; ISSN: 0031-8655
 DT Journal
 LA English
 AB The reaction of tetrakis(dimethylamino)ethylene (I) with O in non-OH solvents gives tetramethylurea, tetramethyloxamide (II), tetramethylhydrazine, and bis-(dimethylamino)methane in yields that are almost independent of solvent and temperature, or whether chemiluminescence occurs. Autoxidn. in aqueous solution, however, gives octamethyloxamidinium peroxide which hydrolyzes to give II and dimethylamine, and also undergoes demethylation to form tetramethyl 2-(dimethylamino)-2-hydroxy-2-(methylamino)acetamidinium and formate salts. Both pathways of autoxidn. occur in LiCl solns., in MeOH, and in H2O-dioxane mixts.
 IT 6415-17-4, Disilanehexamine, dodecamethyl-
 (preparation of)
 RN 6415-17-4 CAPLUS
 CN Disilanehexamine, dodecamethyl- (7CI, 8CI, 9CI) (CA INDEX NAME)



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